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Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry  
SECTION 4 SHORT ANSWER Answer the following questions in  
the space provided. 1. Match each of the following statements  
with the process(es) to which they apply, using one of the  
choices below: (1) fission only (3) both fission and fusion

## **Chapter 21 Nuclear Chemistry Review Answers**

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maker : August Dunbar. nucleons. protons and neutrons. nuclide.  
An atom identified by the number of protons and neutrons in its  
nucleus. mass defect. The difference between the mass of an  
atom and the sum of the masses of its protons, neutrons, and

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electrons.

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About This Chapter The Nuclear Chemistry chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons of nuclear chemistry. Each of these simple and...

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Chapter 21: Nuclear Chemistry. This is a vocabulary test for Chapter 22: Nuclear Chemistry from the "Modern Chemistry" textbook. STUDY. PLAY. Band of stability. Stable nuclei with favorable neutron-proton ratios. Binding energy per nucleon. 1. The binding energy of the nucleus divided by the number of nucleons it contains 2. High binding energy ...

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A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

### **Answer Key Chapter 21 - Chemistry 2e | OpenStax**

21.7.1 Nuclear Reactors. nuclear reactors the fission is controlled to generate a constant power; reactor core consists of fissionable fuel, control rods, a moderator, and cooling fluid; fission products are extremely radioactive and are thus hard to store; about 20 half-lives needed for products to react acceptable levels for biological exposure

### **21.S: Nuclear Chemistry (Summary) - Chemistry LibreTexts**

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...

Nuclear Chemistry Nuclear Transformations • Rutherford in 1919 performed the first nuclear transformation. • The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes:  ${}_{14}^{21}\text{N} + \text{He} \rightarrow \text{O} + \text{H} + {}_{14}^{17}\text{N}$

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## **Chapter 21 Nuclear Chemistry**

AP Chemistry CHAPTER 21- Nuclear Chemistry 21.1 Radioactivity

- When nuclei change spontaneously, emitting energy, they are said to be radioactive.
- Nuclear chemistry is the study of nuclear reactions and their uses.

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In this chapter, we examine some properties of the atomic nucleus and the changes that can occur in atomic nuclei. Nuclear reactions differ from other chemical processes in one critical

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way: in a nuclear reaction, the identities of the elements change.

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