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How much carbon dioxide gas, in grams, is in a 1.0-L ...

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13.78 kPa 6. Find the partial pressure of carbon dioxide in a gas mixture with a total pressure of 30.4 kPa if the partial pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa.

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30.4 kPa 16.5 kPa 3.7 kPa 10.2 kPa Solutions Manual Chemistry:
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255 CHAPTER 13 SOLUTIONS MANUAL Assume that the amount
of gas is constant in the following problems 11 A sample of air in
a syringe exerts a pressure of 102 atm at 220°C The APPENDIX D
Problem Bank

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